## College Prep Chemistry of the Earth System

Assignment 7H

30 Points

Concentration of Solutions

For the following problems calculate the Molarity of the solution.

- 1. 2.30mol NaCl in 1.90L solution Convert to Molarity of the Solution
- 2. 4.21mol H<sub>2</sub>SO<sub>4</sub> in 0.83L solution Convert Molarity of the Solution
- 3. 0.92mol CuCO<sub>3</sub> in 134.5mL soln Convert Molarity of the Solution
- 4. 1.39mol Ba(OH)<sub>2</sub> in 392.5mL soln Convert Molarity of the Solution
- 5. 194.5g Al(BrO<sub>3</sub>)<sub>3</sub> in 3.10L solution Find mol of Co(NO<sub>3</sub>)<sub>3</sub>, then convert to Molarity of the Solution
- 6. 34.8g Co(NO<sub>3</sub>)<sub>3</sub> in 0.293L solution Find mol of Co(NO<sub>3</sub>)<sub>3</sub>, then convert to Molarity of the Solution