

Noteset I J - In Class Notes (Part 2)

Solving Density Equations

$$D = \frac{m}{V}$$

Base Equation

$$V \cdot D = \frac{m}{\cancel{V}} \cdot \cancel{V}$$

Mult. V by both sides

$$m = D \cdot V$$

Solved for mass

Variables in all eqns.
can be solved for any
other variable

$$\frac{m}{D} = \frac{\cancel{D} \cdot V}{\cancel{D}}$$

divide D by both sides

$$V = \frac{m}{D}$$

Solved for volume

Density Equation Set

$$D = \frac{m}{V}$$

$$m = D \cdot V$$

$$V = \frac{m}{D}$$

D = Density (g/mL) m = Mass (g) V = Volume (mL)

Sample Problems

Ex 1: m = 152.6g
 D = 1.20 g/mL
 Find Volume (mL)

$$V = \frac{m}{D}$$

$$V = \frac{152.6g}{1.20g/mL}$$

$$\underline{V = 127.17mL}$$

Don't forget the unit!

Ex2 : $V = 12.8 \text{ mL}$
 $D = 0.621 \text{ g/mL}$
Find mass (g)

$$m = D \cdot V$$
$$m = 0.621 \text{ g/mL} \cdot 12.8 \text{ mL}$$
$$\underline{\underline{m = 7.95 \text{ g}}}$$

Ex3

$$m = 0.916 \text{ g}$$
$$V = 2.93 \text{ mL}$$

Find density (g/mL)

$$D = \frac{m}{V}$$
$$D = \frac{0.916 \text{ g}}{2.93 \text{ mL}}$$
$$\underline{\underline{D = 0.31 \text{ g/mL}}}$$