

College Prep Chemistry of the Earth System

Assignment 2I

20 Points

Calorimetry

Solve the following calorimetry problems

1. $c_{\text{water}} = 4.184\text{J/g}^\circ\text{C}$, $m = 100.0\text{g}$, $T_{\text{final}} = 89.3^\circ\text{C}$, $T_{\text{ini}} = 100.0^\circ\text{C}$, $q_{\text{water}} = \underline{\hspace{1cm}}\text{J}$

$$q_{\text{metal}} = \underline{\hspace{1cm}}\text{J} \quad (q_{\text{metal}} = -q_{\text{water}})$$

2. $c_{\text{water}} = 4.184\text{J/g}^\circ\text{C}$, $m = 100.0\text{g}$, $T_{\text{final}} = 69.2^\circ\text{C}$, $T_{\text{ini}} = 100.0^\circ\text{C}$, $q_{\text{water}} = \underline{\hspace{1cm}}\text{J}$

$$q_{\text{metal}} = \underline{\hspace{1cm}}\text{J} \quad (q_{\text{metal}} = -q_{\text{water}})$$

3. $q_{\text{food}} = -4923.6\text{J}$, $q_{\text{water}} = \underline{\hspace{1cm}}\text{J}$
($q_{\text{water}} = -q_{\text{food}}$)

$$m_{\text{water}} = 100.0\text{g}, \quad C_{\text{water}} = 4.184\text{J/g}^\circ\text{C}$$
$$\Delta T = \underline{\hspace{1cm}}^\circ\text{C}$$

4. $q_{\text{food}} = -9215.2\text{J}$, $q_{\text{water}} = \underline{\hspace{1cm}}\text{J}$
($q_{\text{water}} = -q_{\text{food}}$)

$$m_{\text{water}} = 100.0\text{g}, \quad C_{\text{water}} = 4.184\text{J/g}^\circ\text{C}$$
$$\Delta T = \underline{\hspace{1cm}}^\circ\text{C}$$