

Name \_\_\_\_\_ Period \_\_\_\_\_

CP Chemistry of the Earth

Assignment 4H – Atomic Ion Nomenclature Review

20 Points

Define the following terms based on the ion review

The following is a chart showing the group and valance electrons w/ Lewis Dot Structures

Group Number	1A 1	2A 2	1B – 10B	3A 13	4A 14	5A 15	6A 16	7A 17	8A 18
Valance Electrons (e <sup>-</sup> )	1	2	2	3	4	5	6	7	8
Ion Charge	+1	+2	Var.	+3	+4	-3	-2	-1	0
Lose/Gain e <sup>-</sup>	Lose 1	Lose 2	Var.	Lose 3	Lose 4	Gain 3	Gain 2	Gain 1	Gain 0

Complete the chart below using the attached ion data sheet

Ion	Ion Name*	Ion Charge**	Cation/Anion	Binary/Poly Ion	Lose/Gain e <sup>-</sup>
Na <sup>1+</sup>	Sodium Ion	+1	Cation	Binary	Lose 1e <sup>-</sup>
Ba <sup>2+</sup>					
N <sup>3-</sup>					
Cl <sup>1-</sup>					
NO <sub>3</sub> <sup>1-</sup>					
CO <sub>3</sub> <sup>2-</sup>	Carbonate Ion	-2	Anion	Polyatomic	Gain 2e <sup>-</sup>
BrO <sub>3</sub> <sup>1-</sup>					
CrO <sub>4</sub> <sup>2-</sup>					

\*See Ion Chart \*\*Charge = Charge on Ion Given

Cation = + ion, Anion = - ion Binary = X<sup>1+</sup> (-ide ion), Polyatomic = BrO<sub>3</sub><sup>1-</sup> (-ate ion, -ite ion)

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