| Name | | Period | |
|--|------------------------------|---|-----------|
| College Prep Chemistry | | | |
| Assignment 4D – Introduced Define the following term | uction to Chemical Reactions | ons | 20 Points |
| | | | |
| Chemical Reaction | | Yields (→) | |
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| | | | |
| Reactants (of reaction) | | Products (of reaction) | |
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| For the following reaction | ons list the reactants and p | products | |
| 2Na + Cl₂ → 2NaCl | | $Al_2O_3 \bullet 3H_2O \Rightarrow Al_2O_3 + 3H_2O$ | |
| Reactants | Products | Reactants | Products |
| | | | |
| | | | |
| $2H_2O \rightarrow 2H_2 + O_2$ | | $Ca + Zn_2S \rightarrow 2Zn + CaS$ | |
| Reactants | Products | Reactants | Products |
| | | | |
| | | | |
| $3AgBr + Cu(NO_3)_3 \rightarrow 3AgNO_3 + CuBr_3$ | | $3HF + Al(OH)_3 \rightarrow 3H_2O + AlF_3$ | |
| Reactants | Products | Reactants | Products |
| | | | |
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