

Lab 2 – Density of a Solid and a Liquid**40 Points****Introduction**

In this lab we will be solving for density, the mass (weight) of an object based on the object's volume (*size*). To solve for density we need to measure the volume and mass of the sample using basic laboratory techniques.

Density is found using the density equation:

$$D = m / v, \text{ where } D = \text{Density (g/mL)}, m = \text{mass (g)}, \text{ and } v = \text{volume (mL)}$$

Volume by Water Displacement

The volume of a solid can also be found using the displacement method. To use the displacement method the initial volume (*initial, without solid*) and final volume (*with solid*) using the equation below:

$$V_{\text{solid}} = V_{\text{after}} - V_{\text{start}}$$

Density Tables for Metals

Element	Aluminum	Brass	Copper	Gold	Iron	Lead	Nickle
Density	2.60g/mL	8.50g/mL	8.96g/mL	19.3g/mL	7.87g/mL	11.3g/mL	8.90g/mL
Element	Platinum	Silver	Steel	Tin	Tungsten	Vanadium	Zinc
Density	21.5g/mL	10.5g/mL	7.86g/mL	7.26g/mL	4.51g/mL	6.0g/mL	7.14g/mL

Lab Procedure**Density of a Solid****Volume by Displacement**

1. Add water to the halfway mark on a graduated cylinder (*record volume*)
2. Add metal and record new volume (*record volume*)
3. Subtract new volume from original volume
4. Use volume and mass above to solve for density
5. Determine metal based on chart above/properties

Data Table with Initial Calculations (V_{metal} and Radius)

Metal	V_{start} (mL)	V_{after} (mL)	V_{metal} ($V_{\text{after}} - V_{\text{start}}$)	Mass (g)
Silver (Shiny) (Dark)				
Silver (Not Shiny) (Light)				
Gold	20mL	29mL	9mL	71.6g

$29\text{mL} - 20\text{mL} \rightarrow 9\text{mL}$

Calculations

Density (Solid Measurements)

$$D = \text{Mass (m)} / V_{\text{metal}}$$

$$D = \frac{M_{\text{metal}}}{V_{\text{metal}}}$$

Silver (Shiny)
D = _____
D = _____

Silver (Not Shiny)
D = _____
D = _____

Gold
D = $\frac{71.6 \text{ g}}{9 \text{ mL}}$
D = 7.96 g/mL

Results

What were the metals for each metal cylinder?

Compare results from each density calculations to chart on front of lab document

Cylinder	Silver (Shiny)	Silver (Not Shiny)	Gold
Metal			Brass

(use color + density)