Period

## Lab 2 – Density of a Solid and a Liquid

**40 Points** 

#### Introduction

In this lab we will be solving for density, the mass (weight) of an object based on the object's volume (*size*). To solve for density we need to measure the volume and mass of the sample using basic laboratory techniques.

Density is found using the density equation:

$$D = m / v$$
, where  $D = Density (g/mL)$ ,  $m = mass (g)$ , and  $v = volume (mL)$ 

### Volume by Water Displacement

The volume of a solid can also be found using the displacement method. To use the displacement method the initial volume (*initial*, *without solid*) and final volume (*with solid*) using the equation below:

$$V_{solid} = V_{after} - V_{start}$$

## Density Tables for Metals

Element	Aluminum	Brass	Copper	Gold	Iron	Lead	Nickle
Density	2.60g/mL	8.50g/mL	8.96g/mL	19.3g/mL	7.87g/mL	11.3g/mL	8.90g/mL
Element	Platinum	Silver	Steel	Tin	Tungsten	Vanadium	Zinc
Density	21.5g/mL	10.5g/mL	7.86g/mL	7.26g/mL	4.51g/mL	6.0g/mL	7.14g/mL

#### Lab Procedure

Density of a Solid

Volume by Displacement

- 1. Add water to the halfway mark on a graduated cylinder (record volume)
- 2. Add metal and record new volume (record volume)
- 3. Subtract new volume from original volume
- 4. Use volume and mass above to solve for density
- 5. Determine metal based on chart above/properties

#### Data Table with Initial Calculations ( $V_{metal}$ and Radius)

Metal	$egin{array}{c} V_{ ext{start}} \ (ml) \end{array}$	$rac{ m V_{after}}{(ml)}$	$V_{ ext{metal}} \ (V_{ ext{after}} - V_{ ext{start}})$	Mass (g)
Silver (Shiny)	35.0mL	43.lmL		59.1g
Silver (Not Shiny)	35.0mL	42.9mL		20.3g
Gold	35.0mL	42.9mL		79.28g

Dark Silver Light Silver yellow ovange

$\sim$ 1	1 1	•	
(∵al	cm	latı	ons

Density (Solid Measurements)

D = Mass	(m)/	$V_{\text{metal}}$
----------	------	--------------------

	(South Filedistil Cilitatis)
	Silver (Shiny)
D -	
D = -	
D=	

Silver (	Not Shiny)
D =	
D -	
D =	

	Gold
D = -	
D-	
D=	

# Results

What were the metals for each metal cylinder?

Compare results from each density calculations to chart on front of lab document

Cylinder	Silver (Shiny)	Silver (Not Shiny)	Gold
Metal			