CP Chemistry of the Earth Assignment 2I – Binary Ionic Compounds of Transition Elements Complete the following ionic compounds based on the chart below using the cross method. Elements	Name Period Period							
Elements Zn1+ Cl Ni2+ P Pb4+ N			Compound	a of Transition 1	Elements		20 Points	
Elements						ng the cross me		
Ion Charge								
Charge	Elements	Zn^{1+}	C1	Ni ²⁺	P	Pb ⁴⁺	N	
Charge	Ion							
Cross Method Me								
Method								
Method								
Conic Formula								
Formula	Method							
Formula								
Formula								
Elements Ca O Cu³+ N W6+ S								
Cross Method Elements Sn ⁴⁺ O Ag ¹⁺ F Cr ³⁺ Se Cross Method Ionic Charge Cross Method Ionic Charge	Formula							
Cross Method Elements Sn ⁴⁺ O Ag ¹⁺ F Cr ³⁺ Se Cross Method Ionic Charge Cross Method Ionic Charge				<u> </u>				
Cross Method Ionic Formula Elements Cross Method Ag1+ F Cross Method Ionic Charge Cross Method Ionic Cross Method Ionic	Elements	Ca	O	Cu ³⁺	N	W^{6+}	S	
Cross Method Ionic Formula Elements Cross Method Ag1+ F Cross Method Ionic Charge Cross Method Ionic Cross Method Ionic	Ion							
Cross Method Ionic Formula Elements Sn ⁴⁺ O Ag ¹⁺ F Cr ³⁺ Se Cross Method Ionic Charge								
Method Ionic Formula Elements Sn ⁴⁺ O Ag ¹⁺ F Cr ³⁺ Se Cross Method Ionic								
Method Ionic Formula Elements Sn ⁴⁺ O Ag ¹⁺ F Cr ³⁺ Se Cross Method Ionic								
Ionic Formula Elements Sn ⁴⁺ O Ag ¹⁺ F Cr ³⁺ Se Ion Charge Cross Method Ionic	Cross							
Elements Sn ⁴⁺ O Ag ¹⁺ F Cr ³⁺ Se Ion Charge Cross Method Ionic	Method							
Elements Sn ⁴⁺ O Ag ¹⁺ F Cr ³⁺ Se Ion Charge Cross Method Ionic								
Elements Sn ⁴⁺ O Ag ¹⁺ F Cr ³⁺ Se Ion Charge Cross Method Ionic								
Elements Sn ⁴⁺ O Ag ¹⁺ F Cr ³⁺ Se Cross Method Ionic								
Ion Charge Cross Method Ionic	Formula							
Ion Charge Cross Method Ionic				1				
Cross Method Ionic	Elements	Sn^{4+}	O	Ag^{1+}	F	Cr ³⁺	Se	
Cross Method Ionic	Ion							
Cross Method Ionic								
Method Ionic								
Method Ionic								
Ionic	Cross							
	Method							
rormuia								
	Formula							

Period _____